This article was downloaded by: On: 23 January 2011 Access details: Access Details: Free Access Publisher Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Journal of Carbohydrate Chemistry

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713617200

Addition Reactions of Glycal Esters: Access to Glycosyl Donors of Kdo, dglycero-d-talo- and d-glycero-d-galacto-2-Octulosonic Acid Residues Paul Kosma^a; Harold Sekljic^a; Gregor Balint^a

^a Institut für Chemie, Universität für Bodenkultur, Wien, Austria

To cite this Article Kosma, Paul , Sekljic, Harold and Balint, Gregor(1996) 'Addition Reactions of Glycal Esters: Access to Glycosyl Donors of Kdo, d-*glycero*-d-*talo*- and d-*glycero*-d-*galacto*-2-Octulosonic Acid Residues', Journal of Carbohydrate Chemistry, 15: 6, 701 – 714

To link to this Article: DOI: 10.1080/07328309608005686 URL: http://dx.doi.org/10.1080/07328309608005686

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

ADDITION REACTIONS OF GLYCAL ESTERS: ACCESS TO GLYCOSYL DONORS OF Kdo, D-glycero-D-talo- AND D-glycero-

D-galacto-2-OCTULOSONIC ACID RESIDUES

Paul Kosma,* Harald Sekljic and Gregor Balint

Institut für Chemie, Universität für Bodenkultur, Gregor-Mendel-Str. 33, A-1180 Wien, Austria

Received November 27, 1995 - Final Form May 6, 1996

ABSTRACT

Addition reactions of O-acetylated glycal esters of Kdo mono-, α -(2 \rightarrow 8)- and α -(2 \rightarrow 4)- linked Kdo disaccharide derivatives **1a** - **c** with NIS in acetic acid afforded good yields of *trans*-diaxial as well as minor amounts of *trans*-diequatorial and *cis*-configured 2-O-acetyl-3-deoxy-3-iodo derivatives, which were efficiently reduced with Bu₃SnH/AIBN to give the corresponding per-O-acetylated Kdo methyl ester derivatives. Similar reactions of **1a** with NBS or NCS furnished the *trans*-diaxial 2-O-acetyl-3-bromo-3-deoxy- as well as 3-chloro-3-deoxy derivatives as the main products. Reaction of **1a** with NBS in aqueous MeCN provided the 2,3-*trans*-bromohydrin derivative **11c**, which upon treatment with DBU in MeCN gave the elimination product **11** and the α -2,3-anhydro derivative **12** as a suitable donor of glycosides with D-glycero-D-talo- or D-glycero-D-galacto configuration, respectively.

INTRODUCTION

Functionalization of 2-deoxy-2,3-dehydro derivatives of N-acetylneuraminic acid methyl ester derivatives has been thoroughly studied in the context of developing novel stereospecific glycosyl donors.¹ Related reactions of 2-deoxy-2,3-dehydro compounds of

Downloaded At: 08:31 23 January 2011

Kdo, which are frequently formed in substantial amounts during glycosylation reactions with Kdo donors such as halides² or thioglycosides,³ have received less attention. Addition of phenylsulfenyl chloride to glycal ester derivatives of Kdo⁴ afforded 2 α -chloro-3 β -phenylthio derivatives, which could be stereospecifically converted into glycosides of α -anomeric configuration due to the presence of a stereocontrolling auxiliary at C-3. In a similar fashion, phenylselenyl triflate or chloride was efficiently employed in glycosylation procedures.⁵ Addition of bromine was reported to give a *trans*-diaxial 2,3-dibromo compound with excellent α -selectivity and sufficient reactivity towards primary alcohols.⁶ In addition, reaction of 1a with 3-chloroperbenzoic acid was reported to afford a 2,3-anhydro derivative, which was transformed into ketosides of D-glycero-D-talo- as well as D-glycero-D-galacto configuration,⁷ respectively. The assignment of the configuration of the oxirane ring, however, was tentatively based on a comparison with NMR data of 1,2-anhydro derivatives of D-mannose and D-glucose, respectively.

For these reasons we have set out to investigate addition reactions of glycal esters of Kdo with the intention to reuse these compounds for subsequent glycosylation reactions and furthermore, to devise an unambiguous structural proof for the configurational assignment of 2,3-anhydro-Ko derivatives.

RESULTS AND DISCUSSION

Acetoxyiodination⁸ (NIS/acetic acid/60 °C) of the glycal methyl ester derivative 1a afforded the 2,3-*trans*-diaxial addition product 3a as the major isomer in 71% yield together with the 2,3-*trans*-diequatorial isomer 4a (14%) and a small proportion of the 2,3-*cis* product 2a (7%), which were all separated by silica gel chromatography (Scheme 1). The preferential formation of the thermodynamically favored diaxial isomer is consistent with previous findings on related reactions of *N*-acetylneuraminic acid derivatives.¹ The configuration of C-3 was readily deduced from the ¹H NMR data indicating a *trans*-arrangement of H-3 and H-4 for the D-glycero-D-galacto-derivatives ($J_{3,4}$ = 11.6 and 11.9 Hz for 4a and 2a) and a *cis*-relationship for the D-glycero-D-talo-isomer ($J_{3,4}$ = 5.0 Hz for 3a). The anomeric configuration of the isomers was established following reduction of 3a with AIBN/Bu₃SnH in toluene, which gave the known⁹ α -linked 2-O-acetyl derivative 5a in 76% yield.



Similarly, the disaccharide derivatives 1b and 1c, originating as byproducts in the synthesis of *Chlamydia*-specific and enterobacterial Kdo antigens,^{2b,10} were transformed without cleavage of the ketosidic bonds into the 3-deoxy-3-iodo compounds 2b - 4b and 3c - 4c, again with preferential formation of the 2,3-*trans*-diaxial isomers 3b and 3c. Reduction of 3b and 3c with AIBN/Bu₃SnH in toluene proceeded smoothly to give 5b and 5c, which may be further used in glycosylation reactions following conversion into the corresponding disaccharide bromide derivatives. Thus, the sequence of acetoxyio-dination/reduction provides an efficient way to reuse these glycal esters and should also be applicable for NeuAc-2-en derivatives.

Addition reactions with NBS or NCS in neat acetic acid proved less advantageous, since they required higher temperatures and gave rise to smaller proportions of the 2,3*trans*-diaxial adducts (Scheme 2). The assignment of the configuration of C-3 was again based on the large values of $J_{3,4}$ for the isomers **7a**, **8a,b** and **9a,b**, whereas the smaller values of $J_{3,4}$ for **10a,b** were consistent with the D-glycero-D-talo-configuration (Table).



8 a, b

9 a, b

a: X = Ci **b**: X = Br

<u></u>	X= I			X= Br		X= Cl		
	2a	3a	4a	9b	10b	8a	9a	10a
H-3	4.33	4.51	4.39	4.27	4.40	4.28	4.25	4.38
H-4	5.34	5.02	5.36	5.37	5.46	5.34	5.34	5.51
H-5	5.34	5.45	5.34	5.43	5.40	5.48	5.46	5.40
H-6	4.15	4.24	5.21	4.16	4.24	5.24	4.16	4.22
H-7	5.24	5.34	5.12	5.24	5.37	5.11	5.24	5.36
H-8a	4.41	4.52	4.39	4.42	4.52	4.39	4.43	4.53
H-8b	4.06	4.15	4.13	4.06	4.17	4.13	4.07	4.17
J _{3,4}	11.9	5.0	11.6	11.2	4.5	11.5	11.2	4.1
J _{3,5}	-	0.8	-	-	1.0	-	-	0.9
$J_{4,5}$	n.d. ^b	3.3	3.1	3.3	3.6	3.2	3.3	3.7
$J_{5,6}$	1.3	1.9	1.4	1.2	1.9	1.5	1.5	1.9
J _{6,7}	9.9	9.9	9.7	9.9	9.9	9. 8	9.9	9.9
$J_{7,86}$	2.3	2.2	2.1	2.2	2.4	2.2	2.3	2.3
$J_{7,8b}$	3.7	3.2	4.7	3.6	3.2	4.8	3.6	3.2
$J_{\rm 8a,8b}$	-12.5	-12.5	-12.3	-12.5	-12.5	-12.3	-12.5	-12.5

Table 1. ¹H NMR chemical shifts (δ , ppm) and coupling constants (*J*, Hz) for NXS-adducts^a.

* Spectra were recorded in CDCl₃ at 295 K. ^b not determined.

Furthermore, in case of the NCS addition reaction, chloro-substitution at the anomeric center was observed. The α -anomeric configuration of the resulting 2,3-dichloroderivative 7a was determined using selective proton decoupled ¹³C NMR measurements developed by Haverkamp *et al.* and Hori *et al.*¹² for derivatives of *N*-acetylneuraminic acid derivatives. Thus, selective decoupling of the ¹H NMR signal of the methyl ester group allowed the determination of long range coupling constants $J_{Cl, H3ax}$. The ¹³C NMR signals of C-1 at 166.5, 164.0 and 163.6 ppm (for compounds **8a**, **9a** and **10a**) appeared as a doublet for **8a** (3.4 Hz) and singlets for **9a** and **10a**, respectively. Accordingly. the C-1 signal of 7a was measured as a singlet at 163.6 ppm. Interestingly, H-6 of the β configured 2-*O*-acetyl derivatives **4a**, **4b** and **8a** experienced a pronounced downfield shift in the ¹H NMR spectra (δ 5.00 - 5.44, Table 1); a similar effect may be noted in related 3-deoxy-3-halogeno derivatives of *N*-acetylneuraminic acid⁹.

Recently, glycosides of an octulosonic acid (Ko) being isosteric to Kdo except for an additional, axially oriented OH-group at C-3 have been reported as constituents of the inner core in lipopolysaccharides of *Acinetobacter calcoaceticus* NCTC 10305 and *Pseudomonas cepacia*.¹³ The configuration of this octulosonic acid was determined as Dglycero-D-talo by ¹H and ¹³C NMR measurements and GC-MS comparison with synthetic model compounds of D-glycero-D-talo and D-glycero-D-galacto configuration, respectively. The synthetic derivatives were prepared from a 2,3-anhydro compound obtained via reaction of 1a with 3-chloroperbenzoic acid. By comparison with NMR data from 1,2anhydro derivatives of D-mannose and D-glucose, respectively, the anomeric configuration of the oxirane ring was deduced as β -D-glycero-D-talo.⁷

For the unambiguous configurational assignment of the 2,3-anhydro compound 12, the glycal ester 1a was transformed into the unstable bromohydrin derivatives 9c and 10c in 19% and 76% yield, respectively (Scheme 3). Whereas treatment of 10c with N,Nethyldiisopropylamine in either MeCN or CH_2Cl_2 did not give substantial amounts of the 2,3-anhydro derivative 12, reaction with 1,8-diazabicyclo-[5.4.0.]-undec-7-ene (DBU) in MeCN at 0 °C afforded 12 in 30% yield together with the crystalline 3,4-unsaturated bromohydrin derivative 11 in 46% yield. The structure of 11 was deduced from the ¹H NMR data, which showed a singlet exchangeable with D₂O, a doublet at 6.55 ppm coupled to H-5, whereas H-6,7,8a and 8b were similar to O-acetylated Kdo derivatives. The spectroscopic and physical data of 12 were in full agreement with the previously reported values of the 2,3-anhydro derivative obtained by direct epoxidation in the presence of mCPBA.⁷ Therefore, the published configurational assignment of the oxirane ring has accordingly to be revised as α -D-glycero-D-galacto.



Scheme 3

EXPERIMENTAL

General methods. Melting points were determined with a Kofler hot stage and are uncorrected. Optical rotations were measured with a Perkin-Elmer 243 B polarimeter. ¹H NMR spectra were recorded with a Bruker AC 300F instrument and tetramethylsilane as the internal standard; coupling constants are first order. Homonuclear 2D NMR spectroscopy was performed with Bruker standard software. Thin-layer chromatography was performed on Merck precoated plates (5x10 cm, layer thickness 0.25 mm, Silica Gel $60F_{254}$); spots were detected by spraying with anisaldehyde-H₂SO₄ reagent. Column chromatography was performed on Merck Lichroprep columns (size A, 24x1; B, 31x2.5 and C, 44x3.7 cm; silica gel 40-63 μ m) under pressure (0.2 MPa). Elemental analyses were performed by Dr. J. Theiner, Mikroanalytisches Laboratorium am Institut für Physikalische Chemie, Universität Wien. Methyl 2,4,5,7,8-Penta-O-acetyl-3-deoxy-3-iodo- α -D-glycero-D-galacto-2-octulopyranosonate (2a), Methyl 2,4,5,7,8-Penta-O-acetyl-3-deoxy-3-iodo- α -D-glycero-D-talo-2-octulopyranosonate (3a), and Methyl 2,4,5,7,8-Penta-O-acetyl-3-deoxy-3-iodo-B-Dglycero-D-galacto-2-octulopyranosonate (4a). A solution of 1a (0.24 mmol, 100 mg) and N-iodosuccinimide (0.67 mmol, 150 mg) in acetic acid (20 mL) was heated at 60 °C for 15h. The reaction mixture was then poured into ice-cold saturated aq. NaHCO₃ (200 mL) and extracted with CH₂Cl₂ (100 mL). The organic layer was washed with 10% aq. Na₂S₂O₃ / water, dried (Na₂SO₄) and concentrated. Purification of the residue on a column of silica gel (B, 3:1 toluene-EtOAc) afforded first a mixture of 4a and 2a, then 3a as the major product (100 mg, 71%), colorless crystals, mp 133-134 °C (EtOAc-hexane); [α]_D +52° (*c* 1.0, CHCl₃). ¹H NMR (CDCl₃): δ 1.98, 2.03, 2.09, 2.14, 2.17 (5s, 15H, CH₃CO), 3.86 (s, 3H, CO₂CH₃).

Anal. Calcd for C₁₉H₂₅IO₁₃: C, 38.79; H, 4.28. Found: C, 38.59; H, 4.14.

The mixture of 4a and 2a was rechromatographed (*B*, 5:1 toluene-EtOAc), which furnished 4a (21 mg, 14%) as a syrup, $[\alpha]_D$ +133° (*c* 0.8, CHCl₃). ¹H NMR (CDCl₃): δ 2.00, 2.05, 2.02, 2.12, 2.16 (5s, 15H, CH₃CO), 3.82 (s, 3H, CO₂CH₃).

Anal. Calcd for C₁₉H₂₅IO₁₃: C, 38.79; H, 4.28. Found: C, 39.16; H, 4.23.

Further elution yielded 2a as the slower migrating isomer (11 mg, 7%), colorless crystals, mp 165-168 °C (EtOAc-hexane), $[\alpha]_D$ +111° (c 0.6, CHCl₃). ¹H NMR (CDCl₃): δ 1.98, 2.03, 2.05, 2.14, 2.23 (5s, 15H, CH₃CO), 3.86 (s, 3H, CO₂CH₃).

Anal. Calcd for C₁₉H₂₅IO₁₃: C, 38.79; H, 4.28. Found: C, 38.45; H, 4.22.

O-(Methyl 4,5,7,8-Tetra-*O*-acetyl-3-deoxy-α-D-manno-2-octulopyranosylonate)- $(2\rightarrow 8)$ -(methyl 2,4,5,7-tetra-*O*-acetyl-3-deoxy-3-iodo-α-D-glycero-D-galacto-2-octulopyranosonate) (2b), O-(Methyl 4,5,7,8-Tetra-*O*-acetyl-3-deoxy-α-D-manno-2-octulopyranosylonate)- $(2\rightarrow 8)$ -(methyl 2,4,5,7-tetra-*O*-acetyl-3-deoxy-3-iodo-α-D-glycero-D-talo-2octulopyranosonate) (3b) and O-(Methyl 4,5,7,8-Tetra-O-acetyl-3-deoxy-α-D-manno-2octulopyranosylonate)- $(2\rightarrow 8)$ -(methyl 2,4,5,7-tetra-O-acetyl-3-deoxy-3-iodo-β-D-glycero-Dgalacto-2-octulopyranosonate) (4b). Reaction conditions (1b : 0.54 mmol, 410 mg; NIS: 1.2 mmol, 0.27 g) and work-up were as described for 1a. Column chromatography (C, 3:2 toluene-EtOAc) gave first a (1:1) mixture of 2b and 4b (55 mg, 11%) as a syrup. ¹H NMR (CDCl₃): δ 1.96, 2.00, 2.04, 2.05, 2.06, 2.08, 2.11, 2.15 (8s, 24H, CH₃CO), 2.07 (t, 1H, H- 3'a), 2.20 (dd, 1H, $J_{3'e,3'a} = -12.5$, $J_{3'e,4'} = 5.0$ Hz, H-3'e), 3.58 (dd, 1H, $J_{8a,8b} = -11.2$, $J_{8b,7} = 3.7$ Hz, H-8b), 3.65 (dd, 1H, $J_{8a,7} = 2.3$ Hz, H-8a), 3.77 (s, 3H, CO₂CH₃), 3.90 (s, 3H, CO₂CH₃), 4.09 (dd, 1H, $J_{6,5'} = 1.4$, $J_{6,7'} = 9.5$ Hz, H-6'), 4.24 (dd, 1H, $J_{8'b,8'a} = -12.1$, $J_{8'b,7'} = 6.2$ Hz, H-8'b), 4.28 (dd, 1H, H-6 of **2b**), 4.34 (d, 1H, $J_{3,4} = 11.9$ Hz, H-3), 4.49 (dd, 1H, $J_{8'a,7'} = 2.4$ Hz, H-8'a), 4.97 (ddd, 1H, H-7), 5.26 -5.40 (m, 5H, H-4,5,4',5',7'), 5.44 (dd, 1H, $J_{6,5} = 1.3$, $J_{6,7'} = 9.9$ Hz, H-6 of **4b**).

Further elution of the column gave **3b** (355 mg, 66%) as a syrup. $[\alpha]_D$ +45° (c 0.9, CHCl₃). ¹H NMR (CDCl₃): δ 1.99 (t, 1H, $J_{3'a,3'a} = -12.6$ Hz, H-3'a), 1.96, 2.02, 2.04, 2.09, 2.12, 2.13, 2.24 (7s, 24H, CH₃CO), 2.23 (dd, 1H, H-3'e), 3.66 (dd, 1H, $J_{8a,8b} = -11.8$, $J_{8b,7} = 2.7$ Hz, H-8b), 3.77 (s, 3H, CO₂CH₃), 3.77 (dd, 1H, $J_{8a,7} = 2.5$ Hz, H-8a), 3.83 (s, 3H, CO₂CH₃), 4.05 (dd, 1H, $J_{6',5'} = 1.4$, $J_{6',7} = 9.5$ Hz, H-6'), 4.15 (dd, 1H, $J_{7,8b} = 5.1$, $J_{8'a,8'b} = -12.2$ Hz, H-8'b), 4.34 (dd, 1H, $J_{6,5} = 2.0$, $J_{6,7} = 9.9$ Hz, H-6), 4.51 (dd, 1H, $^4J_{3,5} = 0.8$, $J_{3,4} = 4.9$ Hz, H-3), 4.55 (dd, 1H, $J_{7,8'a} = 2.4$ Hz, H-8'a), 5.04 (dd, 1H, $J_{4,5} = 3.6$ Hz, H-4), 5.12 (ddd, 1H, $J_{4',5'} = 3.0$, $J_{4',3'a} = 5.0$, $J_{4',3'a} = 12.3$ Hz, H-4'), 5.15 (dt, 1H, H-7), 5.24 (ddd, 1H, H-7'), 5.34 (br. d, 1H, H-5'), 5.55 (ddd, 1H, H-5).

Anal. Calcd for $C_{34}H_{45}IO_{23}$: C, 43.05; H, 4.78. Found: C, 42.83; H, 4.65.

4,5,7,8-Tetra-O-acetyl-3-deoxy-Q-D-manno-2-octulopyranosylonate)-O-(Methyl $(2 \rightarrow 4)$ -(methyl 2,5,7,8-tetra-O-acetyl-3-deoxy-3-iodo-Q-D-glycero-D-talo-2-octulopyranosonate) (3c) and O-(Methyl 4,5,7,8-Tetra-O-acetyl-3-deoxy-Q-D-manno-2octulopyranosylonate)-(2-++)-(methyl 2,5,7,8-tetra-O-acetyl-3-deoxy-3-iodo-B-D-glycero-Dgalacto-2-octulopyranosonate) (4c). Reaction conditions (1c : 0.09 mmol, 70 mg NIS : 0.44 mmol, 0.1 g) and work-up were as described for 1a. Column chromatography (B, 1): 1 toluene-EtOAc) gave first the faster migrating isomer 4c, then 3c. Yield for 4c: 4.9 mg (5%), colorless syrup; ¹H NMR (CDCl₃): δ 1.98, 2.00, 2.04, 2.05, 2.07 (double intensity), 2.11, 2.15 (7s, 24H, CH₃CO), 2.10 - 2.19 (m, 2H, H-3'a, H-3'e), 3.58 and 3.84 (2s, 6H, CO_2CH_3 , 4.09 (dd, 1H, $J_{8_{8,8b}} = -12.1$, $J_{8b,7} = 3.7$ Hz, H-8b), 4.17 (dd, 1H, $J_{8b,7} = 7.2$, $J_{8b,8'a}$ =-11.9 Hz, H-8'b), 4.39 (dd, 1H, J_{847} =1.8 Hz, H-8a), 4.56 (dd, 1H, J_{43} =11.6, J_{45} =2.5 Hz, H-4), 4.60 (dd, 1H, $J_{7.8}$ =2.9 Hz, H-8'a), 4.67 (d, 1H, H-3), 4.85 - 4.90 (m, 2H, H-6,7), 5.01 (dd, 1H, $J_{6,7}$ =9.3, $J_{6,5}$ =1.8 Hz, H-6'), 5.25 (dd, 1H, $J_{5,6}$ =1.0 Hz, H-5), 5.28 (ddd, 1H, H-7'), 5.40 (ddd, 1H, H-5'), 5.50 (ddd, 1H, $J_{4'5'} = 2.8, J_{4'3'} = 6.0, J_{4'3'} = 11.5$ Hz, H-4'). Yield for 3c: 68 mg (78 %), colorless syrup. $[\alpha]_D$ +84° (c 1.1, CHCl₃). ¹H NMR (CDCl₃): δ 1.97, 1.99, 2.01, 2.05, 2.08, 2.13, 2.15, 2.17 (8s, 24H, CH₃CO), 2.00 - 2.17 (m, 1H, H-3'e), 2.24 (t, 1H, $J_{3'a,3'e} = J_{3'a,4'} = 11.6$ Hz, H-3'a), 3.86 and 3.89 (2s, 6H, CO₂CH₃), 3.99 (dd, 1H, $J_{3'a,8'b} = -12.2$, $J_{8'b,7'} = 3.5$ Hz, H-8'b), 4.08 (dd, 1H, $J_{6',5'} = 1.2$ Hz, H-6'), 4.13 (dd, 1H, H-8b), 4.15 (dd, 1H, H-6), 4.38 (dd, 1H, ⁴J_{3,5} =0.5, $J_{3,4} = 5.2$ Hz, H-3), 4.44 (dd, 1H, $J_{4,5} = 3.6$ Hz, H-4), 4.50 (dd, 1H, $J_{8a,8b} = -12.5$, $J_{8a,7} = 2.2$ Hz, H-8a), 4.82 (dd, 1H, $J_{8'a,7'} = 2.8$ Hz, H-8'a), 5.16 (ddd, 1H, $J_{6',7'} = 9.2$ Hz, H-7'), 5.25 (ddd, 1H, $J_{7,8b} = 3.0$, $J_{7,6} = 10.0$ Hz, H-7), 5.30 (ddd, 1H, H-5), 5.41 (ddd, 1H, $J_{4',5'} = 2.5$, $J_{4',3'e} = 4.7$ Hz, H-4'), 5.43 (dd, 1H, H-5').

Anal. Calcd for C₃₄H₄₅IO₂₃: C, 43.05; H, 4.78. Found: C, 42.64; H, 4.56.

Methyl 2,4,5,7,8-Penta-*O*-acetyl-3-deoxy-α-D-*manno*-2-octulopyranosonate (5a). A solution of 3a (0.09 mmol, 51 mg), azabisisobutyronitrile (AIBN, 20 mg) and tributyltin hydride (0.93 mmol, 250 µL) in dry toluene (40 mL) was stirred at 80 °C for 4h under N₂. The solution was concentrated and purified by flash chromatography on silica gel (*B*, 1:1 toluene-EtOAc) giving 5a (30 mg, 76%) as colorless crystals, mp 155-157 °C, $[\alpha]_D$ +95°(*c* 1.2 CHCl₃); (lit:¹⁰ mp 155-158 °C, $[\alpha]_D$ +87.1°). ¹H NMR (CDCl₃): δ 1.99, 2.00, 2.05, 2.11, 2.14 (5s, 15H, CH₃CO), 2.23 (m, 2H, H-3*a*, 3*e*), 3.81 (s, 3H, CO₂CH₃), 4.11 (dd, 1H, *J*_{8b,7} =4.0, *J*_{8b,8a} =-12.5 Hz, H-8b), 4.18 (dd, 1H, *J*_{6,7} =9.5 Hz, H-6), 4.48 (dd, 1H, *J*_{8a,7} =2.5 Hz, H-8a), 5.22 (ddd, 1H, H-7), 5.33 (ddd, 1H, *J*_{4,5} =3.0 Hz, H-4), 5.39 (br. d, 1H, H-5).

Methyl 2,4,5,7,8-Penta-O-acetyl-3-deoxy-6-D-manno-2-octulopyranosonate (6a). A solution of 4a (0.05 mmol, 30.9 mg) in toluene (25 mL) was treated with AIBN (20 mg) and Bu₃SnH (0.1 mL) as described for 5a. Column chromatography (A, 1:1 toluene-EtOAc) afforded 6a as a syrup (22 mg, 90 %), $[\alpha]_D$ +39° (c 0.7, CHCl₃). ¹H NMR (CDCl₃): δ 1.99, 2.00, 2.08, 2.11, 2.12 (5s, 15H, CH₃CO), 2.20 (t, 1H, $J_{3e,3a} = J_{3a,4} = 12.7$ Hz, H-3a), 2.41 (dd, 1H, $J_{3e,4} = 4.7$ Hz, H-3e), 3.79 (s, 3H, CO₂CH₃), 4.20 (dd, 1H, $J_{8b,7} = 4.9$, $J_{8b,8a} = 12.2$ Hz, H-8b), 4.43 (dd, 1H, $J_{8a,7} = 2.5$ Hz, H-8a), 4.63 (dd, 1H, $J_{6,5} = 1.2$, $J_{6,7} = 9.7$ Hz, H-6), 5.16 (ddd, 1H, $J_{4,5} = 2.3$ Hz, H-4), 5.17 (ddd, 1H, H-7), 5.35 (br. d, 1H, H-5).

Anal. Calcd for C₁₉H₂₆O₁₃: C, 49.35; H, 5.67. Found: C, 49.13; H, 5.73.

O-(Methyl 4,5,7,8-Tetra-O-acetyl-3-deoxy-α-D-manno-2-octulopyranosylonate)-(2->8) -(methyl 2,4,5,7-tetra-O-acetyl-3-deoxy-α-D-manno-2-octulopyranosonate) (5b). A solution of 3b (0.328 mmol, 310 mg), AIBN (20 mg) and Bu₃SnH (1.66 mmol, 0.44 mL) in toluene (50 mL) was heated at 80 °C for 4h under N₂. The solution was concentrated, diluted with CH₂Cl₂ (50 mL), washed with saturated aq NaHCO₃, dried (Na₂SO₄) and evaporated. Chromatography of the residue on silica gel (*B*, 1:1 toluene-EtOAc) gave **5b** as a syrup (240 mg, 89%). $[\alpha]_D$ +83° (*c* 1.1, CHCl₃); (lit:¹⁴ $[\alpha]_D$ +90°). ¹H NMR data were identical to literature values.¹⁴

O-(Methyl 4,5,7,8-Tetra-*O*-acetyl-3-deoxy- α -D-manno-2-octulopyranosylonate)-(2->4)-(methyl 2,5,7,8-tetra-*O*-acetyl-3-deoxy- α -D-manno-2-octulopyranosonate) (5c). A solution of 3c (0.06 mmol, 58 mg), AIBN (15 mg) and Bu₃SnH (0.16 mmol, 0.04 mL) in toluene (20 mL) was heated at 80 °C for 6h. Work-up as described for 5b gave 5c (43 mg, 85% yield) as colorless crystals, mp 177-179 °C (EtOAc-hexane), lit:^{11a} mp 178-179 °C. ¹H NMR data were identical to literature values.^{11a}

(4,5,7,8-Tetra-O-acetyl-3-chloro-3-deoxy-o-D-glycero-D-galacto-2-octulo-Methyl pyranosyl chloride) onate (7a), Methyl 2,4,5,7,8-Penta-O-acetyl-3-chloro-3-deoxy-B-Dglycero-D-galacto-2-octulopyranosonate (8a), Methyl 2,4,5,7,8-Penta-O-acetyl-3-chloro-3deoxy-a-D-glycero-D-galacto-2-octulopyranosonate (9a) and Methyl 2,4,5,7,8-Penta-Oacetyl-3-chloro-3-deoxy-o-D-glycero-D-talo-2-octulopyranosonate (10a). A solution of 1a (0.66 mmol, 290 mg) and N-chlorosuccinimide (3 mmol, 400 mg) in glacial acetic acid (50 mL) was stirred overnight at 120 °C. The solution was cooled to room temperature and evaporated. The residue was diluted with water and dichloromethane (50 mL each) and washed with saturated aq. NaHCO₃. The organic layer was dried (Na₂SO₄) and concentrated. Purification of the residue by column chromatography (C, 2:1 toluene-EtOAc) afforded 7a as the higher migrating compound. Yield: 76 mg (16%), colorless syrup, $[\alpha]_{D}$ +141° (c 1.1, CHCl₃). ⁱH NMR (CDCl₃): δ 2.01, 2.05, 2.09 and 2.12 (4s, 12) H, CH₃CO), 3.93 (s, 3H, CO₂Me), 4.09 (dd, 1H, $J_{8b,8a}$ =-12.2, $J_{8b,7}$ =4.3 Hz, H-8b), 4.39 (dd, 1H, J_{8a7} = 2.2 Hz, H-8a), 4.62 (dd, 1H, J_{67} = 9.7, J_{65} = 1.2 Hz, H-6), 4.71 (d, 1H, J_{34} =11.1 Hz, H-3), 5.21 (ddd, 1H, H-7), 5.36 (dd, 1H, J_{45} =3.3 Hz, H-4), 5.49 (dd, 1H, H-5).

Anal. Calcd for $C_{17}H_{22}Cl_2O_{11}$: C, 43.14; H, 4.69. Found: C, 43.51; H, 4.58.

Further elution of the column afforded 8a as a syrup. Yield: 38 mg (8%), $[\alpha]_D$ +63° (c 1.4, CHCl₃). ¹H NMR (CDCl₃): δ 2.00, 2.04, 2.06, 2.12 and 2.17 (5s, 15H, CH₃CO), 3.81 (s, 3H, CO₂Me). Elution of the column with 1:1 toluene-EtOAc gave 9a as colorless crystals, mp 164-166 °C (EtOAc-hexane). Yield : 166 mg (33%), $[\alpha]_D$ +120° (c 1.0,

CHCl₃). ¹H NMR (CDCl₃): δ 1.98, 2.04, 2.05, 2.13 and 2.23 (5s, 15H, CH₃CO), 3.86 (s, 3H, CO₂Me). Finally, **10a** was obtained as the slowest migrating isomer. Yield: 146 mg (29%) of **10a** as syrup; $[\alpha]_D$ +61° (c 1.3, CHCl₃). ¹H NMR (CDCl₃): δ 1.97, 2.03, 2.09 (double intensity) and 2.17 (4s, 15H, CH₃CO), 3.85 (s, CO₂Me).

Anal. Calcd for C₁₉H₂₅ClO₁₃ : C, 45.93; H, 5.07. Found: C, 46.17; H, 5.10.

Methyl 2,4,5,7,8-Penta-O-acetyl-3-bromo-3-deoxy- α -D-glycero-D-galacto-2octulopyranosonate (9b) and Methyl 2,4,5,7,8-Penta-O-acetyl-3-bromo-3-deoxy- α -Dglycero-D-talo-2-octulopyranosonate (10b). A solution of 1a (0.83 mmol, 332 mg) and Nbromosuccinimide (2.16 mmol, 386 mg) in acetic acid (30 mL) was stirred for 15h at 70 °C. Work-up as described for 7a and purification of the residue by column chromatography (C, 3:1 \rightarrow 1:1 toluene-EtOAc) afforded 9b as the higher migrating compound, which was chromatographed a second time using 5:1 toluene-EtOAc. Yield: 181 mg (17%), colorless crystals; mp 173-175 °C (EtOAc-hexane), [α]_D +114° (c 1.0, CHCl₃). ¹H NMR (CDCl₃): δ 1.98, 2.04, 2.06, 2.14 and 2.24 (5s, 15H, CH₃O), 3.86 (s, 3H, CO₂Me).

Anal. Calcd for $C_{19}H_{25}BrO_{13}$: C, 42.16; H, 4.66. Found: C, 42.05; H, 4.58. Further elution gave **10b** as a syrup. Yield: 217 mg (49%); $[\alpha]_D + 51^\circ$ (c 1.1, CHCl₃). ¹H NMR (CDCl₃): δ 1.98, 2.04, 2.10, 2.11 and 2.17 (5s, 15H, CH₃CO), 3.86 (s, 3H, CO₂Me).

Anal. Calcd for C₁₉H₂₅BrO₁₃: C, 42.16; H, 4.66. Found: C, 42.71; H, 4.60.

Methyl 4,5,7,8-Tetra-O-acetyl-3-bromo-3-deoxy- α -D-glycero-D-galacto-2-octulopyranosonate (9c) and Methyl 4,5,7,8-Tetra-O-acetyl-3-bromo-3-deoxy- α -D-glycero-D-talo-2-octulopyranosonate (10c). A solution of 1a (0.447 mmol, 180 mg) and N-bromosuccinimide (0.49 mmol, 110 mg) in 5:1 MeCN-H₂O (6 mL) was stirred at 85 °C for 30 min. The solution was concentrated and subjected to silica gel chromatography (*B*, 1:1 toluene-EtOAc). Pooling of the fractions containing the faster migrating compound gave 9c as a syrup. Yield: 42 mg (19%), $[\alpha]_D$ +75° (*c* 0.9, CHCl₃). ¹H NMR (CDCl₃): δ 2.00, 2.04, 2.08, 2.13 (4s, 12H, CH₃CO), 3.95 (s, 3H, CO₂CH₃), 4.09 (dd, 1H, J_{8b,8a} =-12.3, J_{8b,7} =4.5 Hz, H-8b), 4.38 (dd, 1H, J_{8a,7} =2.5 Hz, H-8a), 4.48 (dd, 1H, J_{6,5} =1.3, J_{6,7} = 9.9 Hz, H-6), 4.55 (m, 1H, H-3), 4.59 (br s, 1H, OH), 5.16 (ddd, 1H, H-7), 5.40 (dd, 1H, J_{4,3} =9.7, J_{4,5} =3.3 Hz, H-4), 5.43 (m, 1H, H-5).

Anal. Calcd for C₁₇H₂₃BrO₁₂: C, 40.90; H, 4.64. Found: C, 41.23; H, 4.59.

Further elution of the column furnished 10c (170 mg, 76%) as a syrup, $[\alpha]_D +11^\circ$ (c 0.9, CHCl₃). ¹H NMR (CDCl₃): δ 2.00, 2.08, 2.09, (3s, 12H, CH₃CO), 3.87 (s, 3H, CO_2CH_3), 4.37 (dd, 1H, $J_{8a,8b} = -12.4$, $J_{8b,7} = 3.1$ Hz, H-8b), 4.40 (dd, 1H, ${}^4J_{3.5} = 0.9$, $J_{3.4} = 4.3$ Hz, H-3), 4.44 (dd, 1H, $J_{6.7} = 10.0$, $J_{6.5} = 1.9$ Hz, H-6), 4.45 (dd, 1H, $J_{8a,7} = 3.5$ Hz, H-8a), 4.58 (s, 1H, OH), 5.36 (dt, 1H, H-7), 5.38 (m, 1H, H-5), 5.51 (dd, 1H, $J_{4.5} = 3.6$ Hz, H-4).

Methyl 5,7,8-Tri-O-acetyl-3-bromo-3,4-dideoxy-D-arabino-oct-3-en-2-ulopyranosonate (11) and Methyl 4,5,7,8-Tetra-O-acetyl-2,3-anhydro- α -D-glycero-D-galacto-2-octulopyranosonate (12). A solution of 10c (0.09 mmol, 45 mg) in dry MeCN (6 mL) was cooled to 0 °C. DBU (5 µL) was added, the solution was stirred for 2 min and immediately applied to column of silica gel (A, 2:1 toluene-EtOAc). Pooling of the faster migrating compound afforded 11 (18 mg, 46%) as colorless prisms, mp 154-155 °C (pentane-EtOAc), [α]_D -59° (c 1.0, CHCl₃). ¹H NMR (CDCl₃): δ 2.03, 2.07, 2.08, (3s, 9H, CH₃CO), 3.93 (s, 3H, CO₂CH₃), 4.25 (dd, 1H, J_{8b,7} =4.4, J_{8b,8a} =-12.4 Hz, H-8b), 4.42 (dd, 1H, J_{8a,7} =2.3 Hz, H-8a), 4.46 (dd, 1H, J_{6,5} =2.4, J_{6,7} =9.8 Hz, H-6), 4.55 (s, 1H, OH), 5.16 (dd, 1H, J₅₄ =6.2 Hz, H-5), 5.27 (ddd, 1H, H-7), 6.55 (d, 1H, H-4).

Anal. Calcd for C₁₅H₁₉BrO₁₀ : C, 41.02; H, 4.36. Found: C, 41.26; H, 4.37.

Further elution gave 12 (12.5 mg, 30%) as a syrup, $[\alpha]_D$ +41° (*c* 0.7, CHCl₃), lit:⁷ +37° (*c* 0.9, CHCl₃). ¹H NMR (CDCl₃): δ 2.02, 2.06, 2.07, 2.10, (4s, 12H, CH₃CO), 3.48 (d, 1H, J_{3,4} =1.5 Hz, H-3), 3.86 (s, 3H, CO₂CH₃), 4.05 (dd, 1H, ⁴J_{6,4} =0.8, J_{6.7} =9.6 Hz, H-6), 4.21 (dd, 1H, J_{8b,8a} = -12.4, J_{8b,7} =4.2 Hz, H-8b), 4.56 (dd, 1H, J_{8a,7} =2.4 Hz, H-8a), 5.20 (dd, 1H, H-7), 5.21 (d, 1H, J_{5,4} =4.3 Hz, H-5), 5.26 (ddd, 1H, H-4). ¹³C NMR (75.47 MHz, CDCl₃): δ 53.39 (OCH₃), 54.70 (C-3), 59.93 (C-4), 61.84 (C-8), 66.44 (C-5), 67.49 (C-7), 68.57 (C-6).

Anal. Calcd for C₁₇H₂₂O₁₂ : C, 48.81; H, 5.30. Found: C, 48.98; H, 5.24.

ACKNOWLEDGMENTS

Technical assistance by Annette Schmincke and measurements of NMR spectra by Andreas Hofinger is gratefully acknowledged. The authors are grateful for financial support by Jubiläumsfonds der Österreichischen Nationalbank (grant N 5427)

REFERENCES

 a) K. Okamoto and T. Goto, *Tetrahedron*, 46, 5835 (1990) and references cited. b) Y. Ito and T. Ogawa, *Tetrahedron Lett.*, 29, 3987 (1988). c) E. Kirchner, F. Thiem, R. Dernick, J. Heukeshoven and J. Thiem, J. Carbohydr. Chem., 7, 453 (1988). d) T. Ercegovic and G. Magnusson, J. Chem. Soc. Chem. Commun., 831 (1994).

- a) H. Paulsen and M. Brenken, Liebigs Ann. Chem., 1113 (1991).
 b) F. W. d'Souza, P. Kosma and H. Brade, Carbohydr. Res., 262, 223 (1994).
- G. J. P. H. Boons, F. L. van Delft, P. A. M. van der Klein, G. A. van der Marel, and J. H. van Boom, *Tetrahedron*, 48, 885 (1992).
- 4. K. Ikeda, S. Akamatsu and K. Achiwa, Chem. Pharm. Bull., 38, 279 (1990).
- 5. K. Ikeda, S. Akamatsu and K. Achiwa, Carbohydr. Res., 189, C1 (1989).
- 6. H. Paulsen, A. Wulff and M. Brenken, Liebigs Ann. Chem., 1127 (1991).
- a) J. Gass, M. Strobl, A. Loibner, P. Kosma and U. Zähringer, *Carbohydr. Res.*, 244, 69 (1993).
 b) P. Kosma, M. Strobl, L. März, S. Kusumoto, K. Fukase, L. Brade and H. Brade, *Carbohydr. Res.*, 238, 93 (1993).
- 8. A. Claesson and K. Luthman, Acta Chem. Scand. B, 36, 719 (1982).
- 9. K. Okamoto, T. Kondo and T. Goto, Bull. Chem. Soc. Jpn., 60, 631 (1987).
- 10. F. M. Unger, D. Stix and G. Schulz, Carbohydr. Res., 80, 191 (1980).
- a) P. Kosma, G. Schulz, F. M. Unger and H. Brade, *Carbohydr. Res.*, 190, 191 (1989).
 b) H. Paulsen and C. Krogmann, *Carbohydr. Res.*, 205, 31 (1990).
- a) J. Haverkamp, T. Spoormaker, L. Dorland, J. F. G. Vliegenthart and R. Schauer, J. Am. Chem. Soc., 101, 4851 (1979). b) H. Hori, T. Nakajima, Y. Nishida, H. Ohrui and H. Meguro, Tetrahedron Lett., 29, 6317 (1988).
- 13. U. Zähringer, H. Moll, P. Kosma and K. Kawahara, Abstracts of Papers, VIIth European Carbohydrate Symposium: Cracow, Poland; August, 1993; B 023.
- 14. P. Kosma, G. Schulz and H. Brade, Carbohydr. Res., 183, 183 (1988).